

Organocatalytic Asymmetric Michael/Hemiketalization/Retro-aldol Reaction of α -Nitroketones with Unsaturated Pyrazolones: Synthesis of 3-Acyloxy Pyrazoles

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Supporting Information

$$R^1$$
 R^2 = Ar R^3 = Ar, HetAr, Alk R^3 R^3 = Ar, HetAr, Alk R^3 R^4 R^4 R^4 R^5 R^4 R^5 R^4 R^5 R^6 R

ABSTRACT: An organocatalytic asymmetric cascade Michael/hemiketalization/retro-aldol reaction between unsaturated pyrazolones and α -nitroketones is described. A bifunctional thiourea catalyst was found to be efficient for this reaction. With 10 mol % of catalyst, high yields as well as excellent enantioselectivities are attained for a variety of 3-acyloxy pyrazoles under mild reaction conditions.

Pyrazoles and pyrazolones are important nitrogen containing heterocylic motifs that are prevalent in a wide range of bioactive compounds having pharmaceutical and agricultural activities. In particular, 3-hydroxypyrazole derivatives that are obtained by aromatization of pyrazolones, have interesting enzyme inhibition and activation properties, and have been broadly used in antidiabetic, and anticancer, and have been broadly used in antidiabetic, and anticancer, and have been broadly used in antidiabetic, and anticancer, and have been broadly used in antidiabetic, and have been broadly used in antidiabetic, and anticancer, and have been broadly used in antidiabetic, and herbicidal studies. For instance, some aryl-substituted 3-(3-(dimethylamino)propyloxy)-1H-pyrazoles display potent activation of soluble guanylate cyclase and potent inhibition of platelet aggregation (Figure 1). Similarly, O-pyrazole

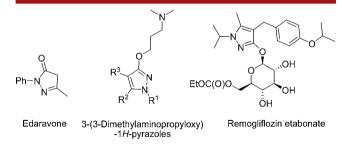


Figure 1. Biologically active 3-hydroxy pyrazole derivatives.

glucopyranoside and galactopyranoside derivatives such as remogliflozin etabonate (Figure 1), ^{2d} are inhibitors of human sodium-glucose contransporters 1 and 2 (SGLT1 and SGLT2) and may be used for the treatment of diabetes. Thus, the development of efficient methods for the enantioselective construction of 3-hydroxy as well as 3-acyloxy pyrazoles having stereogenic centers is important for the discovery of new chiral drugs and other utilities.

In recent years, unsaturated pyrazolones have been exploited as electrophiles in a variety of organocatalytic Michael and cascade reactions. Analogously, pyrazolones have also been found to be suitable nucleophiles in a range of asymmetric reactions. Alao and co-workers first reported the organocatalytic asymmetric synthesis of 3-hydroxy pyrazole via an aza-Michael addition reaction of pyrazolones to α,β -unsaturated ketones. Ma and co-workers have shown one example of the synthesis of 3-hydroxy pyrazole in their development of the organocatlytic Michael reaction to nitroolefins followed by a dearomatization reaction. In contrast, from unsaturated pyrazolones, only a single report for the synthesis of 3-hydroxy pyrazoles has been disclosed by the Wang group (Scheme 1). However, to the best of our knowledge, a direct asymmetric synthesis of 3-acyloxy pyrazoles is still not known. Herein, we

Scheme 1. Organocatalytic Asymmetric Synthesis of 3-Hydroxy/Acyloxy Pyrazoles from Unsaturated Pyrazolones

Previous work: Asymmetric synthesis of 3-hydroxy pyrazoles^{3d}

$$R^{1}$$
 R^{3}
 R^{2}
 R^{3}
 R^{3

This work: Direct asymmetric synthesis of 3-acyloxy pyrazoles

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present a method for the synthesis of 3-acyloxy pyrazoles via a Michael/hemiketalization/retro-aldol strategy employing unsaturated pyrazolones and α -nitroketones.

The investigations were initiated by performing a model reaction between alkylidene pyrazolone 1a and 2-nitro-1-phenylethanone (2a) with quininidine derived bifunctional thiourea catalyst I in dichloromethane solvent at 0 °C (Table 1,

Table 1. Catalyst Screening and Optimization of Reaction Conditions

entry ^a	catalyst	solvent	yield (%) ^b	ee (%) ^c
1	I	CH_2Cl_2	80	-44
2	II	CH_2Cl_2	82	30
3	III	CH_2Cl_2	84	-36
4	IV	CH_2Cl_2	76	36
5	v	CH_2Cl_2	85	61
6	VI	CH_2Cl_2	80	70
7	VII	CH_2Cl_2	85	70
8	VII	CHCl ₃	89	82
9	VII	$(CH_2Cl)_2$	87	87
10	VII	$PhCH_3$	88	90
11	VII	PhCF ₃	91	99

^aReaction condition: 0.05 mmol of **1a** and 0.05 mmol of **2a** in 0.5 mL of solvent using 10 mol % catalyst. ^bIsolated yield after silica gel column chromatography. ^cDetermined by HPLC.

entry 1). Pleasingly a product was isolated in 80% yield and was identified to be 3aa by ^1H and ^{13}C NMR analysis. Previously, only a related kind of Michael-benzoyl transfer reaction was reported in the reaction of α -nitroketone with α , β -unsaturated- α -ketoesters. The enantioselectivity of 3aa was not improved using hydroquinine derived thiourea catalyst II (Table 1, entry 2). Similar kinds of enantioselectivities were achieved with Takemoto catalyst III and squaramide catalyst IV (Table 1, entries 3–4). A higher enantioselectivity was achieved with catalyst V having an Indane moiety (Table 1, entry 5). Then tert-leucine derived bifunctional thiourea catalysts VI and VII were examined. These catalysts were found to be efficient, and in particular, catalyst VII having a piperidine motif provided the product 3aa in 85% yield with 70% ee (Table 1, entry 7). Then the effect of solvent on the yield and enantioselectivity was

studied, and delightfully promising results were attained. For example, enhancements in enantioselectivities were observed using chloroform and 1,2-dichloroethane solvent (Table 1, entries 8–9). Nonpolar solvent such as toluene was also efficient and afforded the product 3aa in 90% ee (Table 1, entry 10). Finally the best solvent was found to be α , α , α -trifluorotoluene, and the product 3aa was isolated in 91% yield with 99% ee (Table 1, entry 11).

After finding the optimal conditions we examined the scope of the reaction. Initially a variety of pyrazolones 1 having different benzylidene substitutents were tested (Table 2). It was

Table 2. Scope of Pyrazolones with Varied Benzylidene Substituents

O2N-

R^1	0 NO ₂		alyst VII mol %)	\mathbb{R}^1	
N N O	T Ph		F ₃ , 0 °C	N _N OCOPh Ph 3aa-ma	
1a-m	2a		3a		
entry ^a	\mathbb{R}^1	3	yield (%) ^b	ee (%) ^c	
1	Ph	3aa	91	99	
2	$4-MeC_6H_4$	3ba	86	97	
3	4-OMeC ₆ H ₄	3ca	85	97	
4	4 - ${}^{t}BuC_{6}H_{4}$	3da	89	85	
5	$4-FC_6H_4$	3ea	65	97	
6	4-ClC ₆ H ₄	3fa	93	98	
7	4 -Br C_6H_4	3ga	93	94	
8	$3-MeC_6H_4$	3ha	88	93	
9	3-OMeC ₆ H ₄	3ia	83	88	
10	3 -Br C_6H_4	3ja	57	88	
11	$2\text{-MeC}_6\text{H}_4$	3ka	93	95	
12	$2\text{-FC}_6\text{H}_4$	3la	80	89	
13	$2,4-Me_2C_6H_3$	3ma	94	94	

^aReactions were carried out with 0.1 mmol of 1 and 0.1 mmol of 2a in α,α,α -trifluorotoluene at 0 °C for 5 days. ^bIsolated yield after silica gel column chromatography. ^cDetermined by HPLC.

found that a range of electron-withdrawing and -donating groups can be embedded in the *ortho-, meta-,* and *para-*position of the aryl group, leading to the synthesis of pyrazoles 3aa—3la in excellent yields and enantioselectivities. A disubstituted aryl group was also tolerated in the reaction, and product 3ma was obtained in excellent enantioselectivity (Table 2, entry 13).

The generality of the reaction was further established by engaging pyrazolones 1 with varied N-substitutions (Table 3). Accordingly, a variety pyrazolones 1n-r with different N-substitutions were prepared and employed in the reaction. To our delight, the reactions progressed well irrespective of the electronic nature of the aryl group and the products 3na-ra were attained in excellent enantioselectivities (Table 3).

The next phase of experiments included screening different α -nitroketones 1 using this method (Scheme 2). As shown in Scheme 2, a wide range of aryl group containing α -nitroketones 2 could be employed in the reaction, and excellent results were achieved. Initially, different *ortho-*, *meta-*, and *para-*substitutions on the phenyl group were incorporated, and delightfully excellent enantioselectivities (ee = 86–96%) were obtained (3ab–ak). A heteroaromatic nitroketone 2l and disubstituted aryl group containing nitroketone 2m also underwent reactions with pyrazolone 1a delivering products 3al and 3am in excellent enantioselectivities. Finally, α -nitroketone 2n having a

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Table 3. Scope of Pyrazolones with Varied N-Substituents

Ph N N O + R ² 1n-r	O Ph NO ₂	(10 PhC	alyst VII mol %) F ₃ , 0 °C days	O ₂ N Ph N OCOPh R ² 3na-ra
entry ^a	\mathbb{R}^2	3	yield (%) ^b	ee (%) ^c
1	4-MeC ₆ H ₄	3na	82	96
2	4-ClC ₆ H ₄	3oa	83	96
3	4-BrC ₆ H ₄	3pa	81	92
4	4-CNC ₆ H ₄	3qa	82	89
5	$2-MeC_6H_4$	3ra	93	99

^aReactions were carried out with 0.1 mmol of 1 and 0.1 mmol of 2a in α , α , α -trifluorotoluene at 0 °C for 5 days. ^bIsolated yield after silica gel column chromatography. ^cDetermined by HPLC.

Scheme 2. Scope of α -Nitroketones^{a,b}

$$\begin{array}{c} \text{Catalyst VII} \\ \text{N} \\ \text{N} \\ \text{O} \\ \text{Ph} \\ \text{1a} \\ \text{2} \\ \\ \text{3ab, R}^3 = 4\text{-MeC}_6\text{H}_4 \ (93\%, \text{ ee} = 96\%) \\ \text{3ac, R}^3 = 4\text{-OMeC}_6\text{H}_4 \ (79\%, \text{ ee} = 94\%) \\ \text{3ad, R}^3 = 4\text{-Pr}_6\text{H}_4 \ (94\%, \text{ ee} = 96\%) \\ \text{3ad, R}^3 = 4\text{-Pr}_6\text{H}_4 \ (94\%, \text{ ee} = 96\%) \\ \text{3af, R}^3 = 4\text{-ClC}_6\text{H}_4 \ (86\%, \text{ ee} = 90\%) \\ \text{Ph} \\ \text{3ag, R}^3 = 4\text{-Br}_6\text{H}_4 \ (90\%, \text{ ee} = 98\%) \\ \text{3ah, R}^3 = 3\text{-OMeC}_6\text{H}_4 \ (93\%, \text{ ee} = 91\%) \\ \text{3aj, R}^3 = 2\text{-MeC}_6\text{H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3aj, R}^3 = 2\text{-MeC}_6\text{H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3aj, R}^3 = 3\text{-OMeC}_6\text{H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3aj, R}^3 = 3\text{-OMeC}_6\text{H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3aj, R}^3 = 3\text{-MeC}_6\text{H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3aj, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-MeC}_6\text{-H}_4 \ (77\%, \text{ ee} = 88\%) \\ \text{3an, R}^3 = 3\text{-$$

^aReactions were carried out with 0.1 mmol of **1a** and 0.1 mmol of **2** in α , α , α -trifluorotoluene at 0 °C for 5 days. ^bIsolated yield after silica gel column chromatography and ee was determined by HPLC.

3ak, $R^3 = 2\text{-FC}_6H_4$ (81%, ee = 86%)

3al, R³ = 2-thienyl (85%, ee = 95%)

cyclohexyl moiety was screened, and gratifyingly excellent enantioselectivity was maintained.

To demonstrate the synthetic utility of our method, few reactions were carried out on 3aa (Scheme 3). Nickel

Scheme 3. Synthetic Transformations of 3aa

chloride—sodium borohydride treatment followed by a reaction with Boc anhydride resulted in the formation of compound 4 in moderate yield while preserving the excellent enantioselectivity. A similar reaction with nickel chloride—sodium borohydride and benzoyl chloride provided amide 5 in an acceptable yield although a slight erosion in enantioselectivity was detected.

The absolute configuration of the product 3af was determined to be (S) by X-ray crystallography. The absolute configuration of other products was expected to be the same by analogy. Based on the absolute configuration a plausible mechanism has been depicted in Scheme 4. It is expected

Scheme 4. Proposed Mechanism

that nitroketone **2a** is activated by the piperidine moiety of the catalyst **VII** whereas a thiourea motif binds with pyrazolone **1a**. Since the *Re* face of **1a** is blocked by catalyst **VII**, addition takes place from the *Si* face to provide intermediate **A**. Then **A** is converted to **B** via hemiketalization. Finally a retro-aldol reaction of **B** delivers product **3aa**.

In conclusion, we have developed a mild and operationally simple Michael—hemiketalization—retro-aldol reaction between unsaturated pyrazolones and α -nitroketones. This reaction furnished diverse 3-acyloxypyrazoles in high yields and with excellent enantioselectivities. Given the high pharmaceutical and agricultural importance of 3-alkoxypyrazoles, our method will be useful for the rapid preparation of these compounds.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03823.

Experimental procedures, characterization data of all the products (\mbox{PDF})

Crystallograhic data for compound 3af (CIF)

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60%, ee = 91%

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Notes

The authors declare no competing financial interest.

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